# Synthesis of sugar-modified analogs of bredinin (mizoribine), a clinically useful immunosuppressant, by a novel photochemical imidazole ring-cleavage reaction as the key step $\dagger^{1}$ 

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#### Abstract

The imidazole nucleoside bredinin (mizoribine) is a clinically useful immunosuppressant. Derivatization of bredinin by the usual nucleoside chemistry is often troublesome due to the unusual zwitterionic structure of the base moiety. We achieve the synthesis of $5^{\prime}$-modified analogs of bredinin via a novel photochemical imidazole ringcleavage reaction as the key step. When a solution of $2^{\prime}, 3^{\prime}-O$-isopropylidenebredinin $\mathbf{5}$ in 0.1 M AcOH is irradiated with a high-pressure mercury lamp, an imidazole ring-cleavage reaction occurs to give the 2 -aminomalonamide riboside derivative $\mathbf{1 6}$ in $71 \%$ yield. Appropriate modifications of the $5^{\prime}$-position of $\mathbf{1 6}$ and subsequent condensation with $(\mathrm{EtO})_{3} \mathrm{CH}$ to reconstruct the imidazole base moiety follow. Using this imidazole ring-cleavage-reconstruction strategy, some biologically important $5^{\prime}$-modified bredinin analogs, i.e., the $5^{\prime}$-phosphate $\mathbf{2}$, the $5^{\prime}$-deoxy derivative $\mathbf{3}$, and the $5^{\prime}-O$-aminopropylcarbamate $\mathbf{4}$ are efficiently synthesized.


## Introduction

Bredinin (mizoribine, 1) is an imidazole nucleoside antibiotic isolated from Eupenicillium brefeldianu ${ }^{2}$ and now clinically used as an immunosuppressant, ${ }^{3,4}$ especially for the transplantation of viscera ${ }^{3 a, b}$ and for autoimmune diseases such as rheumatism. ${ }^{3 c}$ Bredinin as well as its analogs also have antitumor effects in experimental tumor models, ${ }^{5}$ and recently its significant antiviral effect was also reported. ${ }^{6}$ Therefore, chemical modification studies of bredinin may result in the development of useful compounds with efficient pharmacological effects. However, derivatization of bredinin by the conventional methods used in nucleoside chemistry is often troublesome, ${ }^{5 a, 7}$ probably because of the unusual zwitterionic structure of the base moiety. Thus, despite its biological interest, only a few studies on the synthesis of derivatives of bredinin have been reported. ${ }^{5}$
We planned to synthesize the currently biologically important $5^{\prime}$-modified derivatives of bredinin, the $5^{\prime}$-phosphate 2, the $5^{\prime}$-deoxy derivative 3 , and the $5^{\prime}-O-(3$-aminopropyl)carbamate 4 (Fig. 1). However, our first attempt at synthesizing the target compounds from $2^{\prime}, 3^{\prime}-O$-isopropylidenebredinin $\mathbf{5}$, which was readily obtained by treating bredinin with $\mathrm{TsOH}-$ acetone, was unsuccessful. When an electron-withdrawing group was introduced at the $5^{\prime}$-position, intramolecular attack by the 2 -oxygen of the base moiety quickly occurred. For example, treatment of 5 under the usual mesylation conditions or phosphotriester method produced none of the desired $5^{\prime}-\mathrm{O}$-mesylester 6 or $5^{\prime}$-phosphate 7. Both reactions instead gave the $5,5^{\prime}$-anhydro derivative 8 as the major product (Scheme 1). We also tried protecting the phenolic hydroxy group of the base moiety of bredinin with acyl and silyl groups, but this also proved unsuccessful.

Previously, we reported the synthesis of bredinin from 5-

[^0]
bredinin (1)


3


2


4

Fig. 1
amino-4-carbamoylimidazol-1-yl- $\beta$-D-ribofuranoside (AICAR, 9) using a novel photoreaction. ${ }^{7}$ When an acidic aqueous solution of $\mathbf{9}$ or its triacetate $\mathbf{1 0}$ was irradiated with a UV lamp, an imidazole ring-cleavage reaction occurred to give the 2-aminomalonamide derivative $\mathbf{1 1}$ or $\mathbf{1 2}$, respectively as the major product. We found that upon heating with $(\mathrm{EtO})_{3} \mathrm{CH}$ in DMF , compounds $\mathbf{1 1}$ and $\mathbf{1 2}$ were converted into bredinin $\mathbf{1}$ and its triacetate 13, respectively, as shown in Scheme 2. ${ }^{7}$ Accordingly, 2 -aminomalonamide ribosides, such as $\mathbf{1 1}$ and $\mathbf{1 2}$, are considered synthetic equivalents of bredinin. Although 11 or its tri-$O$-acetate $\mathbf{1 2}$ might also be useful intermediates for preparing various bredinin derivatives, ${ }^{5 a}$ the yields of $\mathbf{1 1}$ and $\mathbf{1 2}$ from AICAR or tri- $O$-acetyl-AICAR were low (about $30 \%$ ) and were obtained on only a 100 mg scale once. ${ }^{5 a, 7}$

In this paper, we describe an efficient preparation of 12 and the corresponding $2^{\prime}, 3^{\prime}-O$-acetonide 16 (see Scheme 5, below) from bredinin by a photochemical reaction, and conversion of $\mathbf{1 6}$ into the biologically important $5^{\prime}$-modified bredinin

derivatives, i.e., the $5^{\prime}$-phosphate $\mathbf{2}$, the $5^{\prime}$-deoxy derivative $\mathbf{3}$, and the $5^{\prime}-O-(3-$ aminopropyl)carbamate 4 (Fig. 1), via reconstruction of the imidazole base moiety. ${ }^{8}$

## Results and discussion

## Photoreaction of bredinin

Although the reaction mechanism of the imidazole ringcleavage reaction of AICAR is unclear, it is likely that the 2-aminomalonamide ribosides $\mathbf{1 1}$ and $\mathbf{1 2}$ are produced via acidic hydrolysis of the amidinium intermediate $\mathbf{I}$ (Scheme 3). In the acidic hydrolysis of $\mathbf{I}$, two pathways, producing the desired 2-aminomalonamide ribosides $\mathbf{1 1}$ or $\mathbf{1 2}$ and the undesired


2-aminomalonamide 14, respectively, are possible. We expected that the desired 2-aminomalonamide ribosides $\mathbf{1 1}$ or $\mathbf{1 2}$ might be obtained efficiently if the photoreaction was carried out with bredinin 1, or its sugar-protected derivatives, as a substrate, since the acidic hydrolysis process of the amidinium moiety would not be needed in this case. Thus, we examined the reaction with bredinin.

We first investigated the photoreaction with bredinin $2^{\prime}, 3^{\prime}, 5^{\prime}-$ tri-O-acetate $\mathbf{1 3}$ (Scheme 4), which was readily prepared from


Scheme 4
bredinin by the usual method with $\mathrm{Ac}_{2} \mathrm{O}$-pyridine, in order to make the detection and purification of the reaction products easy. When a solution of $\mathbf{1 3}$ in 0.1 M HCl was irradiated with a high-pressure mercury lamp ( 100 W ) under bubbling of argon, the UV-absorption of the solution rapidly faded, and work-up gave $\mathbf{1 2}$ in $79 \%$ yield as a diastereomeric mixture at the 2-position, which was identical with the photoreaction product of tri- $O$-acetyl-AICAR 10 previously reported. ${ }^{7}$
Next, the photoreaction of $2^{\prime}, 3^{\prime}-O$-isopropylidenebredinin 5 (Scheme 5) was carried out in aq. AcOH to avoid hydrolysis of the acetonide moiety. Thus, a solution of 5 in 0.1 M AcOH was irradiated with a high-pressure mercury lamp to give the desired imidazole ring-cleavage product $\mathbf{1 6}$ in $\mathbf{7 1 \%}$ yield as a diastereomeric mixture at the 2 -position, as in the above reaction with the triacetate 13. In this reaction, the $2,5^{\prime}$-cyclized product 18 was also obtained, in $10 \%$ yield, as a by-product, the structure of which was confirmed by ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, HRMS and elemental analytical data. The configuration at the 2-position was assigned as $S$ by an NOE experiment (Scheme 5). When the photoreaction was carried out with a low-pressure instead of a high-pressure mercury lamp ( 60 W ), the imidazole ringcleavage product $\mathbf{1 6}$ was formed in $68 \%$ yield, with none of the cyclized product $\mathbf{1 8}$ produced. The 2 -amino group of the photoproduct $\mathbf{1 6}$ was protected with a Boc group by the usual method for further derivatization.

There is no precedent for such a photochemical ring-cleavage reaction of imidazole derivatives except for our previous results with AICAR. Although the mechanism of the photoreaction is not clear, protonation of the base moiety would significantly facilitate the reaction (the $\mathrm{p} K_{\mathrm{a}}$ of bredinin is $6.75^{2 a}$ ), because

when performed under neutral conditions the photoreaction proceeded only slowly. Formation of the cyclized product $\mathbf{1 8}$ as a by-product, which is the result of an addition of a $5^{\prime}$-hydroxy group at the 2-position, suggested a possible reaction pathway giving 16 via hydration at the 2-position followed by deformylation (Scheme 6). However this mechanism may not be plaus-

ible, since, when the $N^{2}$-formyl derivative $\mathbf{1 5}$, prepared from 12 by treatment with formic acid and DCC, ${ }^{9}$ was subjected to the above acidic photoreaction conditions with a high-pressure mercury lamp, deformylation did not occur and none of the product 12 was obtained (Scheme 4).

This photoreaction of bredinin derivatives is very useful, since more than 5 g of the photoproduct can be readily obtained at once, while only 100 mg of the product was obtained by the reaction with AICAR derivatives. Therefore, we planned to convert the photoproduct into bredinin derivatives of biological importance.

## Synthesis of bredinin 5'-phosphate

The mechanism of action of bredinin as an immunosuppressant has been studied. ${ }^{3,4,10}$ In cells, bredinin is metabolized into the $5^{\prime}$-phosphate 2 (Fig. 1) by adenosine kinase, ${ }^{10}$ which inhibits cellular inosine monophosphate (IMP) dehydrogenase, an essential enzyme in the de novo synthesis of guanine ribonucleotides. Bredinin inhibits T lymphocyte proliferation due to this antimetabolic effect. Accordingly, the 5'-phosphate 2, the active form of bredinin in cells, should be very useful as a tool in pharmacological studies. However, an efficient method for preparing 2 has not been developed, probably because
of bredinin's unusual chemical features; for instance, when bredinin was treated under Yoshikawa's phosphorylation conditions, ${ }^{11}$ which is the most useful method for preparing nucleoside $5^{\prime}$-phosphates, the desired product 2 was obtained in only $3 \%$ yield. ${ }^{12}$ Attempted phosphorylations at the $5^{\prime}$-hydroxy position of $2^{\prime}, 3^{\prime}-O$-isopropylidenebredinin 5 by the phosphotriester method described above, as well as by the phosphoramidite method, by our hand, were also unsuccessful.

We sought to develop a practical method for preparing 2 via the above photoreaction of bredinin. Therefore, we investigated introducing a phosphate unit at the $5^{\prime}$-position of $\mathrm{N}^{2}$-protected photoproduct 17, and found that a phosphoramidite method with $o$-xylylene $N, N$-diethylphosphoramidite (XEPA) ${ }^{13}$ was effective in this system (Scheme 7). Treatment of 17 with XEPA and tetrazole in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, followed by oxidation with aq. $\mathrm{I}_{2}$, gave the corresponding $5^{\prime}$-phosphotriester 19 in $70 \%$ yield. The isopropylidene and Boc groups of 19 were removed simultaneously with $90 \%$ aq. TFA, and the resulting product, without purification, was heated with $(\mathrm{EtO})_{3} \mathrm{CH}$ in DMF at $90^{\circ} \mathrm{C}$ to give the bredinin $5^{\prime}$-phosphate derivative $\mathbf{2 0}$ in $47 \%$ yield from 19. Hydrogenation of 20 with Pd -carbon in MeOH furnished bredinin 5'-phosphate 2, which was isolated as a disodium salt in $89 \%$ yield, after successive treatment with Dowex $50\left(\mathrm{H}^{+}\right)$ and Diaion WK-20 $\left(\mathrm{Na}^{+}\right)$resins.

## Synthesis of 5'-deoxybredinin

4-Carbamoylimidazolium-5-olate 25, the aglycone of bredinin, is known to show potent antitumor effects in vivo stronger than those of bredinin itself. ${ }^{5 c}$ We designed 5'-deoxybredinin 3 as a potential antitumor agent. This compound was expected to release the aglycone 25 efficiently in tumor cells by the action of nucleoside phosphorylase (NP) (Scheme 8), since the activity of this enzyme is significantly increased in tumor cells compared with normal cells. ${ }^{14,15}$

The synthesis of 3 is shown in Scheme 7. The $N^{2}$-(Bocamino)malonamide riboside derivative 17 was successively treated with $\mathrm{MsCl}-$ py and $\mathrm{NaI}-$ methyl ethyl ketone (MEK) to give the corresponding 5'-deoxy-5'-iodo derivative 21 in $90 \%$ yield. The $5^{\prime}$-iodide 21 was then subjected to radical reduction with $\mathrm{Bu}_{3} \mathrm{SnH}$-azoisobutyronitrile (AIBN) in benzene to give the 5'-deoxy derivative 22 quantitatively. Removal of the protecting groups of 22 with $90 \%$ TFA and subsequent reconstruction of the imidazole ring with $(\mathrm{EtO})_{3} \mathrm{CH}$ afforded the target $5^{\prime}$-deoxybredinin $\mathbf{3}$ in $77 \%$ yield.

## Synthesis of bredinin 5'-O-(3-aminopropyl)carbamate

The 5'-O-(3-aminopropyl)carbamate 4 was designed as a bredinin derivative having an aminoalkyl tether, which can bond covalently to other molecules. Such a bredinin derivative would be very useful for biological studies, which include synthesis of haptens for preparing antibodies to bredinin and also preparation of bredinin-attached resins for affinity chromatography.

Compound 17 was successively treated with carbonyl-diimidazole-4-(dimethylamino)pyridine (DMAP) and N -Cbz-propanediamine- $\mathrm{Et}_{3} \mathrm{~N}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give the $5^{\prime}-\mathrm{O}$-carbamate derivative 23 in $91 \%$ yield. Deprotection and reconstruction of the base by the above method gave the bredinin $5^{\prime}-O$-aminopropylcarbamate 24 , which was hydrogenated with $\mathrm{Pd}-\mathrm{C}$ in the presence of HCl in MeOH to afford the target compound 4, as shown in Scheme 7.

## Conclusions

We have found that the imidazole base moiety of bredinin is cleaved easily by irradiation under aqueous acidic conditions to give 2 -aminomalonamide riboside derivatives and that the base moiety of bredinin can be reconstructed when the photoproduct is heated with $(\mathrm{EtO})_{3} \mathrm{CH}$. Using this imidazole ring-


Scheme 7


Scheme 8
cleavage-reconstruction strategy, several biologically important $5^{\prime}$-modified bredinin analogs were efficiently synthesized.

## Experimental

Mps were measured on a Yanagimoto MP-3 micro-melting point apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ NMR spectra were recorded at 100,400 , and $500 \mathrm{MHz}\left({ }^{1} \mathrm{H}\right)$ and at $100 \mathrm{MHz}\left({ }^{13} \mathrm{C}\right)$. Chemical shifts $(\delta)$ and coupling constants $(J)$ are reported in ppm downfield from TMS and in Hz , respectively. Assignments of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR described are based on COSY and/or DEPT spectra. Mass spectra were obtained by fast-atom bombardment (FAB) or chemical ionization (CI). Thin-layer chromatography (TLC) was done on Merck $60 \mathrm{~F}_{254}$ coated plates. Silica gel chromatography was done with Merck silica gel 5715 or 9385. Reactions were carried out under an argon atmosphere.

## $\mathbf{2}^{\prime}, \mathbf{3}^{\prime}$-O-Isopropylidenebredinin 5

A suspension of bredinin $\mathbf{1}(10.4 \mathrm{~g}, 40 \mathrm{mmol})$ and $\mathrm{TsOH} \cdot \mathrm{H}_{2} \mathrm{O}$
$(16.0 \mathrm{~g}, 84.0 \mathrm{mmol})$ in acetone ( 800 mL ) was stirred at room temperature for 2 h , and then the resulting solution was neutralized with $\mathrm{NH}_{4} \mathrm{OH}(28 \%$ aq.). The resulting precipitate was filtered off and washed well with EtOH , and the filtrate was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}, 10: 1\right)$ to give $5(9.63 \mathrm{~g}, 80 \%)$ as a solid; mp 209-212 ${ }^{\circ} \mathrm{C}$ (from MeOH) (Found: C, 48.14; H, 5.82; $\mathrm{N}, 13.86 . \mathrm{C}_{12} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{6}$ requires C, 48.16; $\left.\mathrm{H}, 5.73 ; \mathrm{N}, 14.04 \%\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ;$ DMSO- $d_{6}+\mathrm{D}_{2} \mathrm{O}$ ) $\delta 8.27$ (s, 1 H ), 5.73 (d, $1 \mathrm{H}, J=2.6$ ), 5.16 (dd, $1 \mathrm{H}, J=5.9,2.6$ ), $4.85(\mathrm{dd}, 1 \mathrm{H}, J=5.9$, 2.6), $4.15(\mathrm{~m}, 1 \mathrm{H}), 3.57(\mathrm{~m}, 2 \mathrm{H}), 1.49$ and 1.30 (each s, each 3 H ); MS (FAB, positive) $m / z 300\left(\mathrm{MH}^{+}\right)$.

## 5,5'-Anhydrobredinin 8

A mixture of 5 ( $299 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) and $\mathrm{MsCl}(88 \mu \mathrm{~L}, 1.3 \mathrm{mmol})$ in pyridine ( 8 mL ) was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h . After addition of water $(1.0 \mathrm{~mL})$, the resulting mixture was evaporated, and the residue was partitioned between EtOAc and brine. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated, and purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}, 100: 1\right.$, then $\left.20: 1\right)$ to give $8(138 \mathrm{mg}, 36 \%)$ as a foam; ${ }^{1} \mathrm{H}$ NMR ( $100 \mathrm{MHz} ; \mathrm{CDCl}_{3}+$ $\left.\mathrm{D}_{2} \mathrm{O}\right) \delta 7.19(\mathrm{~s}, 1 \mathrm{H}), 5.76(\mathrm{~s}, 1 \mathrm{H}), 5.07(\mathrm{~d}, 1 \mathrm{H}, J=5.6), 4.77(\mathrm{~d}$, $1 \mathrm{H}, J=5.6), 4.69(\mathrm{~d}, 1 \mathrm{H}, J=2.2), 4.56(\mathrm{dd}, 1 \mathrm{H}, J=12.9,2.2)$, 4.06 (d, $1 \mathrm{H}, J=12.9$ ), 1.54 and 1.53 (each s, each 3 H ); MS (CI) $m / z 282\left(\mathrm{MH}^{+}\right) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\text {max }} 280 \mathrm{~nm}$.

## $\mathbf{2}^{\prime}, \mathbf{3}^{\prime}, \mathbf{5}^{\prime}$-Tri-O-acetylbredinin $\mathbf{1 3}$

A mixture of bredinin $1(5.18 \mathrm{~g}, 20 \mathrm{mmol})$ and $\mathrm{Ac}_{2} \mathrm{O}(9.5 \mathrm{~mL}$, 100 mmol ) in pyridine ( 50 mL ) was stirred at room temperature for 3 h . After MeOH was added, the solvent was evaporated, and the residue was treated with hot $\mathrm{CHCl}_{3}-\mathrm{MeOH}$ to give crystalline 13 ( $5.13 \mathrm{~g}, 67 \%$ ); mp 188-190 ${ }^{\circ} \mathrm{C}$ (Found: C, 47.05;

H, 5.09; $\mathrm{N}, 10.90 . \mathrm{C}_{15} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{9}$ requires C, 46.76; H, 4.97; N , $10.90 \%$ ); ${ }^{1} \mathrm{H}$ NMR ( 100 MHz ; DMSO- $d_{6}$ ) $\delta 8.35$ ( $\mathrm{s}, 1 \mathrm{H}$ ), 6.88 (br s, 2 H ), $5.80(\mathrm{~m}, 2 \mathrm{H}), 5.52(\mathrm{~m}, 1 \mathrm{H}), 4.42-4.11(\mathrm{~m}, 3 \mathrm{H})$, 2.08, 2.07 and 2.03 (each s, each 3 H ); MS (CI) $m / z 385\left(\mathrm{M}^{+}\right)$; UV (MeOH) $\lambda_{\text {max }} 284 \mathrm{~nm}$.

## 2-Amino- N -(2,3,5-tri- O -acetyl- $\boldsymbol{\beta}$-d-ribofuranosyl)malonamide 12

A solution of $\mathbf{1 3}(4.10 \mathrm{~g}, 10.6 \mathrm{mmol})$ in aq. $\mathrm{HCl}(0.1 \mathrm{M} ; 500$ mL ) was irradiated with a high-pressure mercury lamp ( 100 W , Pyrex filter) at room temperature under bubbling of argon for 3.5 h . After bring neutralized with aq. $\mathrm{NaHCO}_{3}(0.8 \mathrm{M})$, the resulting mixture was concentrated to about 50 mL , and then NaCl was added. The resulting solution was extracted with $\mathrm{CHCl}_{3}$ (3 times), and the combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated. The residue was purified by flash column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}, 17: 1\right)$ to give $12(3.14 \mathrm{~g}, 79 \%)$ as a foam, the ${ }^{1} \mathrm{H}$ NMR spectral data of which were identical with those reported previously. ${ }^{7 a}$

## 2-Formamido- N -(2,3,5-tri- O -acetyl- $\boldsymbol{\beta}$-d-ribofuranosyl)malonamide 15

A mixture of DCC ( $413 \mathrm{mg}, 2.0 \mathrm{mmol}$ ) and formic acid ( 151 $\mu \mathrm{L}, 4.0 \mathrm{mmol}$ ) in $\mathrm{CHCl}_{3}(5 \mathrm{~mL})$ was stirred at $0^{\circ} \mathrm{C}$ for 10 min . A solution of $\mathbf{1 2}(357 \mathrm{mg}, 1.0 \mathrm{mmol})$ in pyridine ( 3 mL ) was added, and the resulting mixture was stirred at $0^{\circ} \mathrm{C}$ for 15 min and then at room temperature for 1 h . The resulting white precipitate was filtered off, and the filtrate was evaporated. The residue was partitioned between $\mathrm{CHCl}_{3}$ and brine, and the aqueous layer was extracted with $\mathrm{CHCl}_{3}$ ( 5 times). The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated, and purified by flash column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}\right.$, $30: 1$, then $25: 1$ ) to give $\mathbf{1 5}(187 \mathrm{mg}, 46 \%)$ as a foam: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 8.32$ and 8.30 (each s, each $0.5 \mathrm{H}, \mathrm{CHO}$ ), 8.18 and 8.10 (each d, each $0.5 \mathrm{H}, 1-\mathrm{NH}, J=8.5$ and 8.8 , respectively), 7.65 and 7.61 (each d, each $0.5 \mathrm{H}, \mathrm{NHCHO}$, $J=6.4$ and 6.4 , respectively), $7.00\left(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{CONH}_{2}\right), 6.40$ and 6.31 (each br s, each $\left.0.5 \mathrm{H}, \mathrm{CONH}_{2}\right), 5.68-5.61\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right)$, $5.32-5.26\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-2^{\prime}\right.$ and $\left.-3^{\prime}\right), 5.10$ and 5.07 (each d, each 0.5 $\mathrm{H}, \mathrm{H}-3, J=6.4$ and 6.4 , respectively), $4.30-4.15\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-4^{\prime}\right.$ and $\mathrm{H}_{2}-5^{\prime}$ ), 2.11, 2.10, 2.09 and 2.08 (each s, total $9 \mathrm{H}, \mathrm{Ac}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta 171.01,170.92,170.17,170.01$, $169.97,169.93,168.40,167.72,166.93$ and $166.51\left(\mathrm{COCH}_{3}\right.$ and $\left.\mathrm{CONH}_{2}\right), 166.93$ and $166.51(\mathrm{CHO}), 82.99$ and $82.66\left(\mathrm{C}-1^{\prime}\right)$, 78.80 and 78.75 (C-4'), 73.59, 73.23, 70.81 and 70.70 (C-2' and $\left.-3^{\prime}\right), 63.46$ (C-5'), 56.16 and 55.93 (C-3), 20.76 and 20.52 (acetyl Me ); HRMS (FAB, positive) $404.1292\left(\mathrm{MH}^{+} . \mathrm{C}_{15} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{10}\right.$ requires $m / z, 404.1305$ ).

## 2-Amino- N -(2,3-O-isopropylidene- $\beta$-D-ribofuranosyl)malonamide 16

Reaction using a low-pressure mercury lamp. A solution of 5 $(5.96 \mathrm{~g}, 20.0 \mathrm{mmol})$ in aq. AcOH ( $0.1 \mathrm{M} ; 600 \mathrm{~mL}$ ) was irradiated with a low-pressure mercury lamp ( 60 W , quartz filter) at room temperature under bubbling of argon for 12 h . After being neutralized with aq. $\mathrm{NaHCO}_{3}(0.8 \mathrm{M})$, the resulting mixture was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}, 10: 1\right.$, then $5: 1$ ) to give $16(3.98 \mathrm{~g}, 68 \%)$ as a foam: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; DMSO- $\left.d_{6}+\mathrm{D}_{2} \mathrm{O}\right) \delta 5.44(\mathrm{~d}, 1 \mathrm{H}, J=2.0), 4.71(\mathrm{~d}, 1 \mathrm{H}, J=6.4)$, $4.51(\mathrm{dd}, 1 \mathrm{H}, J=6.4,2.0), 4.04(\mathrm{~m}, 1 \mathrm{H}), 3.81$ and 3.79 (each s, each 0.5 H , disappeared after 1 h from $\mathrm{D}_{2} \mathrm{O}$-addition), 3.53$3.43(\mathrm{~m}, 2 \mathrm{H}), 1.42$ and 1.26 (each s, each 3 H ); MS (FAB, positive) $m / z 290\left(\mathrm{MH}^{+}\right)$. This compound was rather unstable and therefore was immediately used for the next reaction.

Reaction with high-pressure mercury lamp. A solution of 5 $(1.49 \mathrm{~g}, 5.0 \mathrm{mmol})$ in aq. $\mathrm{AcOH}(0.1 \mathrm{M} ; 500 \mathrm{~mL})$ was irradiated
with a high-pressure mercury lamp ( 100 W , Pyrex filter) at room temperature under bubbling of argon for 6 h . The resulting white precipitate was filtered off and washed with water to give $2,5^{\prime}$-cyclized product 18 ( $145 \mathrm{mg}, 10 \%$ ) as crystals, mp 241 $242{ }^{\circ} \mathrm{C}$ (Found: C, $46.81 ; \mathrm{H}, 5.70 ; \mathrm{N}, 13.80 . \mathrm{C}_{12} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{6} \cdot 1 / 2 \mathrm{H}_{2} \mathrm{O}$ requires C, 46.75; H, 5.89; N, 13.63\%); ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; DMSO- $d_{6} / \mathrm{D}_{2} \mathrm{O}$ ) $\delta 5.37$ (s, $1 \mathrm{H}, \mathrm{H}-1^{\prime}$ ), 5.04 (s, $1 \mathrm{H}, \mathrm{H}-2$ ), 4.83 (d, $1 \mathrm{H}, \mathrm{H}-3^{\prime}, J=5.9$ ), $4.75\left(\mathrm{~d}, 1 \mathrm{H}, \mathrm{H}-2^{\prime}, J=5.9\right), 4.48(\mathrm{~m}, 1 \mathrm{H}$, H-4'), 4.11 (dd, $1 \mathrm{H}, \mathrm{H}-5^{\prime} \mathrm{a}, J=13.2,1.0$ ), 3.74 (dd, $1 \mathrm{H}, \mathrm{H}-5^{\prime} \mathrm{b}$, $J=13.2,2.0$ ), 1.40 and 1.28 (each s, each $3 \mathrm{H}, \mathrm{CHMe}$ ); NOE irradiated $\mathrm{H}-1^{\prime}$, observed $\mathrm{H}-2(1.1 \%)$, $\mathrm{H}-2^{\prime}\left(5.8^{\prime} \%\right)$; irradiated H-2, observed H-1' (3.0\%), H-5'b (6.2\%); ${ }^{13}$ C NMR ( 100 MHz , DMSO- $d_{6}$ ) $\delta 169.68$ (C), 165.46 (C), 111.89 (C), 88.63 (CH), $87.76(\mathrm{CH}), 85.92(\mathrm{CH}), 84.55(\mathrm{CH}), 80.96(\mathrm{CH}), 72.33(\mathrm{C})$, $71.94\left(\mathrm{CH}_{2}\right), 25.95\left(\mathrm{CH}_{3}\right), 24.43\left(\mathrm{CH}_{3}\right) ; \mathrm{MS}(\mathrm{CI}) \mathrm{m} / \mathrm{z} 300$ $\left(\mathrm{MH}^{+}\right)$.
The filtrate was neutralized with aq. $\mathrm{NaHCO}_{3}(0.8 \mathrm{M})$ and evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}, 10: 1\right.$ then 5:1) to give $\mathbf{1 6}(1.02 \mathrm{~g}$, $71 \%$ ) as a foam.

## 2-(tert-Butoxycarbonylamino)- $N$-(2,3- $O$-isopropylidene- $\beta$-dribofuranosyl)malonamide 17

A mixture of $16(867 \mathrm{mg}, 3.0 \mathrm{mmol}), \mathrm{Boc}_{2} \mathrm{O}(981 \mathrm{mg}, 4.5$ $\mathrm{mmol})$, and $\mathrm{Et}_{3} \mathrm{~N}(0.63 \mathrm{~mL}, 4.5 \mathrm{mmol})$ in 1,4-dioxane ( 15 mL ) was stirred at room temperature for 2.5 h . The resulting mixture was evaporated, and the residue was partitioned between EtOAc and brine. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated, and purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\right.$ $\mathrm{MeOH}, 20: 1$ ) to give $17(866 \mathrm{mg}, 74 \%)$ as a foam (Found: C, 48.68; $\mathrm{H}, 6.85 ; \mathrm{N}, 10.27 . \mathrm{C}_{16} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{8} \cdot 1 / 3 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 48.60$; H, $7.05 ; \mathrm{N}, 10.63 \%$ ); ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; DMSO- $d_{6}$ ) $\delta 8.61$ (br d, $1 \mathrm{H}, 1-\mathrm{NH}, J=8.3$ ), $7.47,7.45$ and 7.40 (each br s, total 2 H , $\mathrm{CONH}_{2}$ ), 6.63 (br d, $0.5 \mathrm{H}, 2-\mathrm{NH}, J=8.3$ ), 6.61 (br d, 0.5 H , 2-NH, $J=8.8$ ), 5.41 (dd, $0.5 \mathrm{H}, \mathrm{H}^{\prime}{ }^{\prime}, J=1.5,8.3$ ), 5.38 (dd, 0.5 H, H-1', $J=1.9,8.3$ ), $5.19\left(\mathrm{t}, 0.5 \mathrm{H}, 5^{\prime}-\mathrm{OH}, J=4.9\right), 5.12(\mathrm{t}, 0.5$ $\left.\mathrm{H}, 5^{\prime}-\mathrm{OH}, J=5.4\right), 4.70\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3^{\prime}\right), 4.58-4.48(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-2$ and -2'), 4.01 (m, $\left.1 \mathrm{H}, \mathrm{H}-4^{\prime}\right), 3.52-3.37$ (m, $2 \mathrm{H}, \mathrm{H}-5^{\prime}$ ), $1.42-$ $1.26\left(\mathrm{~m}, 15 \mathrm{H}, t-\mathrm{Bu}\right.$ and $\operatorname{Pr}^{\mathrm{i}}$ ); MS (FAB, positive) $m / z 390$ $\left(\mathrm{MH}^{+}\right)$.

## 2-(tert-Butoxycarbonylamino)- N -[3-oxo-5-O-(2,4,3-benzo-dioxaphosphepan-3-yl)-2,3- $O$-isopropylidene- $\beta$-D-ribofuranosyl]malonamide 19

A mixture of $17(78 \mathrm{mg}, 0.20 \mathrm{mmol})$, tetrazole ( $32 \mathrm{mg}, 0.46$ mmol ), and 3-diethylamino-2,4,3-benzodioxaphosphepane XEPA ( $72 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ was stirred at room temperature for 1 h . To the mixture was added a solution of $\mathrm{I}_{2}(120 \mathrm{mg}, 0.47 \mathrm{mmol})$ in aq. THF $(95 \% ; 4 \mathrm{~mL})$, and the whole was stirred at room temperature for 10 min and then quenched with saturated aq. $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$. The resulting mixture was partitioned between $\mathrm{CHCl}_{3}$ and brine. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, evaporated, and purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{EtOAc}, 1: 1\right.$, then $\mathrm{CHCl}_{3}-$ $\mathrm{MeOH}, 50: 1$ ) to give 19 ( $81 \mathrm{mg}, 70 \%$ ) as a foam; ${ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) \delta 8.15-8.04(\mathrm{~m}, 1 \mathrm{H}), 7.40-7.28(\mathrm{~m}, 4 \mathrm{H}), 7.03(\mathrm{br}$ $\mathrm{s}, 0.5 \mathrm{H}), 6.95(\mathrm{br} \mathrm{s}, 0.5 \mathrm{H}), 6.03-5.90(\mathrm{~m}, 2 \mathrm{H}), 5.64(\mathrm{~m}, 1 \mathrm{H})$, $5.31-5.16(\mathrm{~m}, 4 \mathrm{H}), 4.80-4.70(\mathrm{~m}, 3 \mathrm{H}), 4.36(\mathrm{~m}, 1 \mathrm{H}), 4.31-4.21$ $(\mathrm{m}, 2 \mathrm{H}), 1.52-1.32(\mathrm{~m}, 15 \mathrm{H})$; MS (FAB, positive) $\mathrm{m} / \mathrm{z} 572$ $\left(\mathrm{MH}^{+}\right)$.

## 5'-O-(3-Oxo-2,4,3-benzodioxaphosphepan-3-yl)bredinin 20

A solution of $\mathbf{1 9}(1.27 \mathrm{~g}, 3.4 \mathrm{mmol})$ in $90 \%$ aq. TFA was stirred at room temperature for 15 min . After water was added, the solvent was evaporated, and the residue was coevaporated with water. The residue and $\mathrm{Et}_{3} \mathrm{~N}(1.0 \mathrm{~mL})$ were dissolved in MeOH $(15 \mathrm{~mL})$, and then the solvent was evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}\right.$, $10: 1$, then $5: 1$ ) to give a foam $(1.01 \mathrm{~g})$. A mixture of the foam
and $(\mathrm{EtO})_{3} \mathrm{CH}(734 \mu \mathrm{~L}, 4.4 \mathrm{mmol})$ in DMF $(25 \mathrm{~mL})$ was heated at $90^{\circ} \mathrm{C}$ for 15 min . The resulting mixture was evaporated, and the residue was purified by column chromatography $\left(\mathrm{SiO}_{2}\right.$; $\mathrm{CHCl}_{3}-\mathrm{MeOH}, 10: 1$, then $\left.5: 1\right)$ to give $20(487 \mathrm{mg}, 47 \%)$ as a solid; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; DMSO- $d_{6}+\mathrm{D}_{2} \mathrm{O}$ ) $\delta 8.28$ (s, 1 H ), 7.46-7.41 (m, 4 H$), 5.57(\mathrm{~d}, 1 \mathrm{H}, J=4.9), 5.37-5.29(\mathrm{~m}, 2 \mathrm{H})$, $5.11-5.01$ (m, 2 H ), 4.39 (dd, $1 \mathrm{H}, J=4.9,4.9$ ), 4.29 (m, 1 H ), 4.24-4.19 (m, 2 H ), $4.09(\mathrm{~m}, 1 \mathrm{H})$; HRMS (FAB, positive) $442.1036\left(\mathrm{MH}^{+} . \mathrm{C}_{17} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}\right.$ requires $\mathrm{m} / \mathrm{z}$, 442.1015); UV $\left(\mathrm{H}_{2} \mathrm{O}\right) \lambda_{\text {max }} 280 \mathrm{~nm}$.

## Bredinin 5'-phosphate sodium salt 2

A mixture of $20(297 \mathrm{mg}, 0.67 \mathrm{mmol})$ and $\mathrm{Pd}-\mathrm{C}(10 \%, 100 \mathrm{mg})$ in $\mathrm{MeOH}(10 \mathrm{~mL})$ was stirred at room temperature under atmospheric pressure of $\mathrm{H}_{2}$ for 30 min . The catalyst was filtered off, and the filtrate was evaporated. The residue was partitioned between $\mathrm{CHCl}_{3}$ and water, and the aqueous layer was concentrated in vacuo, and then applied to a column of Dowex 50 resin ( $\mathrm{H}^{+}$-form, $1.8 \times 8 \mathrm{~cm}$, packed with water). The column was eluted with water ( 300 mL ), and the appropriate fractions containing 2 were concentrated to about 3 mL in vacuo. The resulting solution was applied to a column of Diaion WK-20 resin $\left(\mathrm{Na}^{+}\right.$-form, $1.8 \times 10 \mathrm{~cm}$, packed with water). The column was eluted with water. The appropriate fractions containing 2 were evaporated, and the residue was freeze-dried to give pure 2 ( 230 mg , sodium salt, $89 \%$ ) as a solid (Found: C, 27.15; H, 4.17; N, 10.04. $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{NaO}_{9} \mathrm{P} \cdot 7 / 3 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 26.81 ; \mathrm{H}, 4.42 ; \mathrm{N}$, $10.42 \%$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}$ ) $\delta 8.38$ ( s, 1 H ), $5.81(\mathrm{~d}, 1 \mathrm{H}$, $J=4.4), 4.49(\mathrm{dd}, 1 \mathrm{H}, J=4.4,4.9), 4.41(\mathrm{dd}, 1 \mathrm{H}, J=2.4,4.9)$, $4.28(\mathrm{~m}, 1 \mathrm{H}), 4.14-4.04(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{31} \mathrm{P}$ NMR ( $\mathrm{D}_{2} \mathrm{O} ; 125 \mathrm{MHz}$, decoupled with $\left.{ }^{1} \mathrm{H}\right) \delta 1.85(\mathrm{~s}) ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}$ ) $\delta 165.03,156.36,127.03,101.32,87.43$ and 84.41 (d, $J=7.5$ ), $75.61,70.58$ and 64.45 ; MS (FAB, positive) $m / z 362\left(\mathrm{MNa}^{+}\right)$; MS (FAB, negative) $m / z 338\left[(\mathrm{M}-\mathrm{H})^{-}\right] ; \mathrm{UV}\left(\mathrm{H}_{2} \mathrm{O}\right) \lambda_{\max } 279$ nm .

## 2-(tert-Butoxycarbonylamino)- N -(5-deoxy-5-iodo-2,3-O-isopropylidene- $\beta$-D-ribofuranosyl)malonamide 21

A mixture of $17(1.17 \mathrm{~g}, 3.0 \mathrm{mmol})$ and $\mathrm{MsCl}(302 \mu \mathrm{~L}, 3.9$ mmol ) in pyridine ( 30 mL ) was stirred at $0^{\circ} \mathrm{C}$ for 1 h . After addition of water, the solvent was evaporated, and the residue was partitioned between $\mathrm{CHCl}_{3}$ and brine. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated. A mixture of the residue and $\mathrm{NaI}(1.12 \mathrm{~g}, 7.5 \mathrm{mmol})$ in MEK ( 30 mL ) was heated under reflux for 1 h . The resulting precipitate was filtered off, and the filtrate was evaporated. The residue was partitioned between $\mathrm{CHCl}_{3}$ and brine, and the organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated. The residue was purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{EtOAc}, 4: 1\right)$ to give $21(1.35 \mathrm{~g}, 90 \%)$ as a solid (Found: C, $38.55 ; \mathrm{H}, 5.28 ; \mathrm{N}, 8.22 . \mathrm{C}_{16} \mathrm{H}_{26} \mathrm{IN}_{3} \mathrm{O}_{7}$ requires C, 38.49 ; H, 5.25; N, $8.42 \%$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $\delta 7.81$ (d, $0.5 \mathrm{H}, J=8.3$ ), 7.76 (d, $0.5 \mathrm{H}, J=6.8$ ), 6.95 and 6.91 (each br s, each 0.5 H ), 6.01 (dd, $1 \mathrm{H}, J=4.9,4.9$ ), 5.95 (br s, $1 \mathrm{H}), 5.59(\mathrm{~m}, 1 \mathrm{H}), 4.76-4.67(\mathrm{~m}, 3 \mathrm{H}), 4.23(\mathrm{~m}, 1 \mathrm{H}), 3.33-3.22$ (m, 2H), 1.53-1.34 (m, 15 H ); MS (FAB, positive) $\mathrm{m} / \mathrm{z} 500$ $\left(\mathrm{MH}^{+}\right)$.

## 2-(tert-Butoxycarbonylamino)- N -(5-deoxy-2,3- O -isopropyl-idene- $\beta$-D-ribofuranosyl)malonamide 22

A mixture of $21(998 \mathrm{mg}, 2.0 \mathrm{mmol}), \mathrm{Bu}_{3} \mathrm{SnH}(810 \mu \mathrm{~L}, 3.0$ $\mathrm{mmol})$, and $\operatorname{AIBN}(100 \mathrm{mg}, 0.61 \mathrm{mmol})$ in benzene ( 30 mL ) was heated under reflux for 2 h . The solvent was evaporated, and the residue was purified by column chromatography $\left(\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\right.$ $\mathrm{MeOH}, 30: 1)$ to give $22(736 \mathrm{mg}, 98 \%)$ as a foam; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; DMSO- $d_{6}$ ) $\delta 8.77-8.68$ (m, 1 H ), 7.41 (br s, 2 H ), 6.63-6.58(m, 1 H), 5.30-5.21 (m, 1 H), 4.70-4.42 (m, 4 H$)$, 4.02-3.98(m, 1 H$), 1.46-1.16(\mathrm{~m}, 15 \mathrm{H}), 0.88(\mathrm{~m}, 3 \mathrm{H})$; MS (FAB, positive) $m / z 374\left(\mathrm{MH}^{+}\right)$.

## 5'-Deoxybredinin 3

Compound 3 ( $385 \mathrm{mg}, 77 \%$ ) as a solid was obtained from $22(495 \mathrm{mg}, 2.0 \mathrm{mmol})$ as described above for the synthesis of 20, after purification by column chromatography ( $\mathrm{SiO}_{2}$; $\mathrm{CHCl}_{3}-\mathrm{MeOH}, 10: 1$ then $5: 1$ ); mp 217-219 ${ }^{\circ} \mathrm{C}$ (from MeOH , decomp.) (Found: C, 44.32; H, 5.52; N, 16.91. $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{5} \cdot 1 /$ 7 MeOH requires C, $44.62 ; \mathrm{H}, 5.51 ; \mathrm{N}, 16.71 \%$ ); ${ }^{1} \mathrm{H}$ NMR ( 400 MHz; DMSO- $d_{6}+\mathrm{D}_{2} \mathrm{O}$ ) $\delta 8.29$ (s, 1 H ), 6.87 (br s, 2 H ), 5.48 (d, $1 \mathrm{H}, J=4.8$ ), 4.34 (dd, $1 \mathrm{H}, J=4.8,5.1$ ), 3.91-3.79 (m, 2 H ), $1.24(\mathrm{~d}, 3 \mathrm{H}, J=6.2)$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}$ ) $\delta 161.71$, 155.21, 124.43, 98.52, 86.05, 79.34, 74.43, 72.98, 18.74; MS (FAB, positive) $m / z 244\left(\mathrm{MH}^{+}\right)$; UV $\lambda_{\text {max }} 278 \mathrm{~nm}\left(\mathrm{H}_{2} \mathrm{O}\right)$.

## N -\{5-O-[3-(Benzyloxycarbonylamino)propylcarbamoyl]-2,3-O-isopropylidene- $\beta$-d-ribofuranosyl\}-3-(tert-butoxycarbonylamino)malonamide 23

A mixture of $\mathbf{1 7}$ ( $97 \mathrm{mg}, 0.25 \mathrm{mmol}$ ), carbonyldiimidazole ( 61 $\mathrm{mg}, 0.50 \mathrm{mmol}$ ) and DMAP ( $67 \mathrm{mg}, 0.55 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(5 \mathrm{~mL})$ was stirred at room temperature for 6 h . To the mixture was added a mixture of 3-(benzyloxycarbonylamino)propylamine hydrochloride ${ }^{16}(257 \mathrm{mg}, 1.0 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(139 \mu \mathrm{~L}$, $1.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$, and the whole was stirred at room temperature for 16 h . After water was added, the resulting mixture was partitioned, and the organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated. The residue was purified by column chromatography ( $\mathrm{SiO}_{2} ; \mathrm{CHCl}_{3}-\mathrm{MeOH}, 80: 1$ ) to give 23 (138 $\mathrm{mg}, 91 \%$ ) as a foam (Found: C, 53.79; H, 6.51; N, 11.00. $\mathrm{C}_{28} \mathrm{H}_{41} \mathrm{~N}_{5} \mathrm{O}_{11}$ requires C, 53.92; H, 6.63; N, 11.23\%); ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz} ; \mathrm{CDCl}_{3}+\mathrm{D}_{2} \mathrm{O}\right) \delta 7.37-7.32(\mathrm{~m}, 5 \mathrm{H}), 5.87(\mathrm{~s}, 0.7 \mathrm{H})$, $5.81(\mathrm{~s}, 0.3 \mathrm{H}), 5.08(\mathrm{~s}, 2 \mathrm{H}), 4.72-4.41(\mathrm{~m}, 5 \mathrm{H}), 3.91(\mathrm{~d}, 0.7 \mathrm{H}$, $J=12.4$ ), 3.83 (d, $0.3 \mathrm{H}, J=11.9$ ), $3.39-3.18$ (m, 4 H ), $1.73-$ $1.31(\mathrm{~m}, 17 \mathrm{H})$; HRMS (FAB, positive) $624.2888\left(\mathrm{MH}^{+}\right.$. $\mathrm{C}_{28} \mathrm{H}_{42} \mathrm{~N}_{5} \mathrm{O}_{11}$ requires $m / z 624.2881$ ).

## 5'-O-[3-(Benzyloxycarbonylamino)propylcarbamoyl]bredinin 24

A solution of $\mathbf{2 3}(249 \mathrm{mg}, 0.40 \mathrm{mmol})$ in $90 \%$ TFA ( 3 mL ) was stirred at room temperature for 10 min , and then the solution was evaporated. After the residue was coevaporated with water and then with EtOH (twice), the residue was dissolved in EtOH $(2 \mathrm{~mL})$ and $\mathrm{Et}_{3} \mathrm{~N}(1 \mathrm{~mL})$, and the solution was evaporated. The residue was coevaporated with EtOH (3 times), and then heated at $40^{\circ} \mathrm{C}$ in vacuo for 3 h . The resulting syrup was heated at $90^{\circ} \mathrm{C}$ with (EtO) $)_{3} \mathrm{CH}(67 \mu \mathrm{~L}, 0.52 \mathrm{mmol})$ in $\mathrm{DMF}(5 \mathrm{~mL})$ for 20 min , and then the solvent was evaporated. The residue was dissolved in $\mathrm{MeOH}(1 \mathrm{~mL})$ and applied to a column of Dowex 50 resin ( $\mathrm{H}^{+}$-form, $2 \times 15 \mathrm{~cm}$, packed with MeOH ). The column was washed with $\mathrm{MeOH}(200 \mathrm{~mL}$ ), and then developed with $50 \% \mathrm{aq} . \mathrm{MeOH}$. Appropriate fractions were evaporated to give $24(132 \mathrm{mg}, 67 \%)$ as a foam (Found: C, 50.26 ; H, 5.61 ; N, 13.84. $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{5} \mathrm{O}_{9} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 50.1 ; \mathrm{H}, 5.81 ; \mathrm{N}$, $13.91 \%$ ); ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; DMSO- $d_{6}+\mathrm{D}_{2} \mathrm{O}$ ) $\delta 8.24$ (s, 1 H , $\mathrm{H}-2), 5.55(\mathrm{~d}, 1 \mathrm{H}, J=5.9), 5.00\left(\mathrm{~s}, 2 \mathrm{H}\right.$, benzyl $\left.\mathrm{CH}_{2}\right), 4.34(\mathrm{dd}$, $1 \mathrm{H}, \mathrm{H}-2^{\prime}, J=5.9,4.9$ ), 4.16 (dd, $1 \mathrm{H}, \mathrm{H}-5^{\prime} \mathrm{a}, J=11.7,2.4$ ), 4.05-3.99 (m, 3 H, H-3', -4', -5'b), 3.00-2.95 (m, $4 \mathrm{H}, \mathrm{NCH}_{2} \times$ 2), $2.98\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right)$; HRMS ( FAB , positive) $494.1884\left(\mathrm{MH}^{+} . \mathrm{C}_{21} \mathrm{H}_{28} \mathrm{~N}_{5} \mathrm{O}_{9}\right.$ requires $\mathrm{m} / \mathrm{z}$ 494.1887); UV $(\mathrm{MeOH}) \lambda_{\text {max }} 283 \mathrm{~nm}$.

## $\mathbf{5}^{\prime}$ - $\boldsymbol{O}$-(3-Aminopropylcarbamoyl)bredinin hydrochloride 4

A mixture of $24(49 \mathrm{mg}, 0.10 \mathrm{mmol})$, $\mathrm{Pd}-\mathrm{C}(10 \% ; 50 \mathrm{mg})$, and $12 \mathrm{M} \mathrm{HCl}(8.8 \mu \mathrm{~L}, 0.105 \mathrm{mmol})$ in $\mathrm{MeOH}(5 \mathrm{~mL})$ was stirred at room temperature under atmospheric pressure of $\mathrm{H}_{2}$ for 30 min . The catalyst was filtered off, and the filtrate was evaporated. The residue was coevaporated with EtOH ( 3 times), and then the residue was treated with EtOH to give $4 \cdot \mathrm{HCl}(34 \mathrm{mg}$, $86 \%$ ) as a solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{D}_{2} \mathrm{O}$ ) $\delta 8.08$ (s, $1 \mathrm{H}, \mathrm{H}-2$ ), 5.78 (d, $1 \mathrm{H}, \mathrm{H}^{\prime} 1^{\prime}, J=3.4$ ), 4.62 (dd, $1 \mathrm{H}, \mathrm{H}-2^{\prime}, J=5.4,3.4$ ),
4.50 (dd, $\left.1 \mathrm{H}, \mathrm{H}-5^{\prime} \mathrm{a}, J=12.2,1.8\right), 4.41$ (dd, $1 \mathrm{H}, \mathrm{H}-3^{\prime}, J=5.4$, 5.4), 4.31-4.25 (m, $\left.2 \mathrm{H}, \mathrm{H}-4^{\prime},-5^{\prime} \mathrm{b}\right), 3.23\left(\mathrm{t}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right.$, $J=6.8), 3.02\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{NCH}_{2}, J=7.8\right), 1.86\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right) ;{ }^{13} \mathrm{C}$ NMR ( 125 MHz ; DMSO- $\left.d_{6}\right) \delta 164.62,158.30$, 155.97, 126.24, 101.46, 87.41, 82.00, 73.64, 70.00, 63.80, 37.56, 37.23, 27.32; HRMS (FAB, positive) $360.1521\left(\mathrm{NH}^{+} . \mathrm{C}_{13} \mathrm{H}_{22^{-}}\right.$ $\mathrm{N}_{5} \mathrm{O}_{7}$ requires $\left.m / z 360.1519\right)$; UV $\lambda_{\text {max }} 279 \mathrm{~nm}\left(\mathrm{H}_{2} \mathrm{O}\right)$.

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## Notes and references

1 This report constitutes Part 202 of our series Nucleosides and Nucleotides. Part 201: T. Umino, N. Minakawa and A. Matsuda, Tetrahedron Lett., 2000, 41, 6419.
2 (a) K. Mizuno, M. Tsujino, M. Takada, M. Hayashi, K. Atsumi, K. Asano and T. Matsuda, J. Antibiot., 1974, 27, 775; (b) H. Yoshioka, K. Nakatsu, M. Hayashi and K. Mizuno, Tetrahedron Lett., 1975, 4031.
3 (a) H. Amemiya and H. Itoh, in Immunosuppressive Drugs: Developments in Anti-rejection Therapy, ed. A. W. Thomson and T. E. Starzl, Edward Arnold, London, 1993, pp. 161-176; (b) S. A. Giruber, Immunol. Rev., 1992, 129, 5; (c) Y. Shiokawa, M. Honma, K. Shichikawa, A. Miyamoto, S. Hirose, T. Nobunaga, H. Mizushima, S. Sugawara, S. Warabi, H. Kondo and K. Ogawa, Jpn. J. Inflammation, 1991, 11, 375; (d) H. Ishikawa, Curr. Med. Chem., 1999, 6, 575.

4 (a) L. A. Turka, J. Dayton, G. Sinclair, C. B. Thompson and B. S. Mitchell, J. Clin. Invest., 1991, 87, 940; (b) J. S. Dayton, L. A. Turka, C. B. Thompson and B. S. Mitchell, Mol. Pharmacol., 1992, 41, 671.
5 (a) S. Shuto, H. Itoh, E. Endo, K. Fukukawa, M. Tsujino and T. Ueda, Chem. Pharm. Bull., 1987, 35, 3523; (b) N. Yoshida, M. Nakamura, M. Fukui, S. Morisada, S. Ogino, M. Inaba, S. Tsukagoshi and Y. Sakurai, Cancer Res., 1983, 43, 5851; (c) M. Fukui, M. Inaba, S. Tsukagoshi and Y. Sakurai, Cancer Res., 1982, 42, 1098.
6 Y. Kosugi, Y. Saito, S. Mori, J. Watanabe, M. Baba and S. Shigeta, Antiviral Chem. Chemother., 1994, 5, 366.
7 (a) K. Fukukawa, S. Shuto, T. Hirano and T. Ueda, Chem. Pharm. Bull., 1986, 34, 3653; (b) K. Fukukawa, S. Shuto, T. Hirano and T. Ueda, Chem. Pharm. Bull., 1984, 32, 1644.

8 A part of this study has been described in a communication: S. Shuto, K. Haramuishi and A. Matsuda, Tetrahedron Lett., 1996, 37, 187.

9 M. Waki and J. Meienhofer, J. Org. Chem., 1977, 42, 2019.
10 H. Koyama and M. Tsuji, Biochem. Pharmacol., 1983, 32, 200.
11 M. Yoshikawa, T. Kato and T. Takenishi, Bull. Chem. Soc. Jpn., 1969, 42, 3505.
12 H . Watababe, unpublished results: when bredinin was subjected to Yoshikawa's procedure, dehydration of the 4-carbamoyl moiety occurred, giving the corresponding 4-cyano derivative as the major product.
13 Y. Watanabe, Y. Komoda, K. Ebisuya and S. Ozaki, Tetrahedron Lett., 1990, 31, 255.
14 G. Weber, M. E. Burt, R. C. Jackson, N. Prajda, M. S. Lui and E. Takeda, Cancer Res., 1983, 43, 1019.

15 A similar activation of $5^{\prime}$-deoxy-5-fluorouridine (doxifluridine), a clinically useful anticancer drug, by nucleoside phosphorylase in tumor cells is known: J.-P. Sommadossi, C. Aubert, J.-P. Cano, J. Gouveia, P. Ribaud and G. Mathe, Cancer Res., 1983, 43, 930.

16 G. J. Atwell and W. A. Denny, Synthesis, 1984, 1032.


[^0]:    $\dagger^{1} \mathrm{H}$ NMR spectral charts of 15, 19, 20, 22 and 4, and $\mathrm{H}-\mathrm{H}$ COSY spectral charts of $\mathbf{1 5}, \mathbf{1 7}, \mathbf{1 8}, 24$ and 4 are available as supplementary data from BLDSC (SUPPL. NO. 57713, 11 pp.) or the RSC Library. See Instructions for Authors available via the RSC web page (http:// www.rsc.org/authors).

